

Climate Change Impacts of the Non-Kyoto Greenhouse Gases and Aerosols

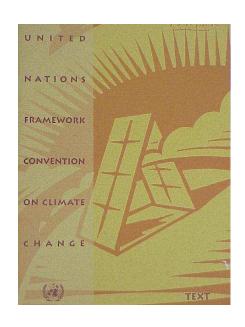
Michael J. Prather Fred Kavli Professor of Earth System Science UC Irvine

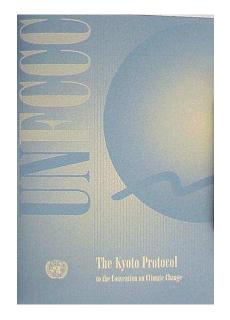


Conventions / Protocols governing Climate Change

1992 United Nations Framework Convention on Climate Change

adopted in 1992 and entered into force in 1994 (188 ratification – 28 Jan 2003)





1997 Kyoto Protocol

ratified 2004, in force 16 Feb 2005

Greenhouse Gases included in the Kyoto Protocol

Annex A

Greenhouse gases

Carbon dioxide (CO₂)

Methane (CH₄)

Nitrous oxide (N₂O)

Hydrofluorocarbons (HFCs)

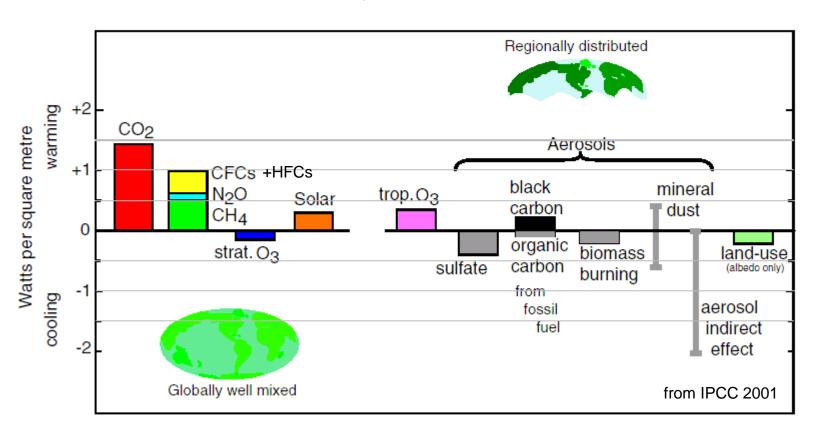
Perfluorocarbons (PFCs)

Sulphur hexafluoride (SF₆)

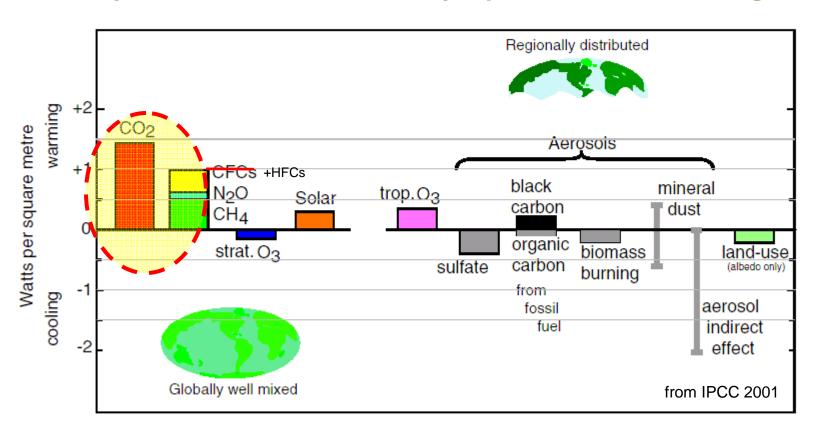
Sectors/source categories

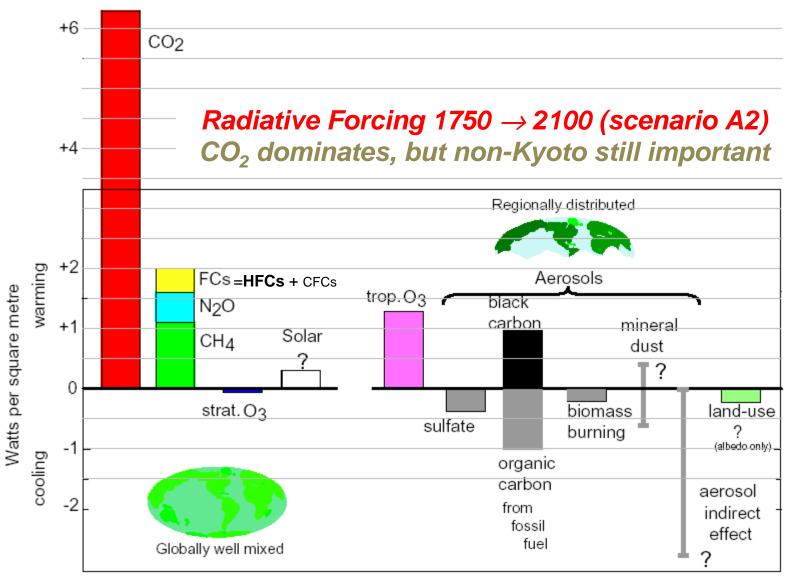
Energy, Fuel combustion, Energy industries, Manufacturing industries and construction, Transport, Other sectors, Fugitive emissions from fuels, Solid fuels, Oil and natural gas, Mineral products, Chemical industry, Metal production, Production of halocarbons and sulphur hexafluoride Consumption of halocarbons and sulphur hexafluoride Agriculture, Enteric fermentation, Manure management Rice cultivation, Agricultural soils, Prescribed burning of savannas Field burning of agricultural residues, Waste, Wastewater handling, Other

Radiative Forcing 1750 \rightarrow 2000 from IPCC TAR 2001, the recent 2007 AR4 is similar



Radiative Forcing 1750 → 2000 Kyoto Greenhouses are only a part of climate forcing





from IPCC 2001

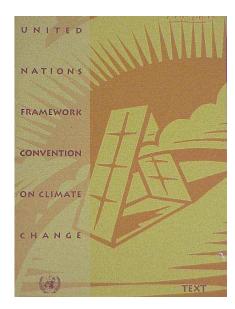
Kyoto Greenhouse Gas Emissions

1990-2002, million metric tons CO₂-equivalent*

Kyoto gas	Global	Annex-Ir	EU-15	USA	CA
FF CO ₂	22811	11447	3308	5423	337
LUCF CO ₂	+7891	-1314	-233	-915	-21
CH ₄	5866	1378	400	629	31
N ₂ O	3254	970	367	420	30
PFCs	92	44	9	15	2.3
HFCs	155	123	40	69	8.6
SF ₆	62	50	12	28	2.5

^{*} Using 1996 IPCC SAR 100-yr GWP.

The Framework Convention is a bit broader in recognizing what forces climate change



UN FCCC ARTICLE 1

"Source" means any process or activity which releases a greenhouse gas, <u>an aerosol or a precursor of a greenhouse gas</u> into the atmosphere.

UN FCCC ARTICLE 3.3

... policies and measures should ... <u>cover all relevant sources</u>, <u>sinks and reservoirs of greenhouse gases</u> ...

Greenhouse Gases and Aerosols not in Kyoto

CFCs & HCFCs (Ozone Depleting Substances, under the Montreal Protocol)

Tropospheric Ozone Precursors (NOx, CO, VOC, CH₄*)

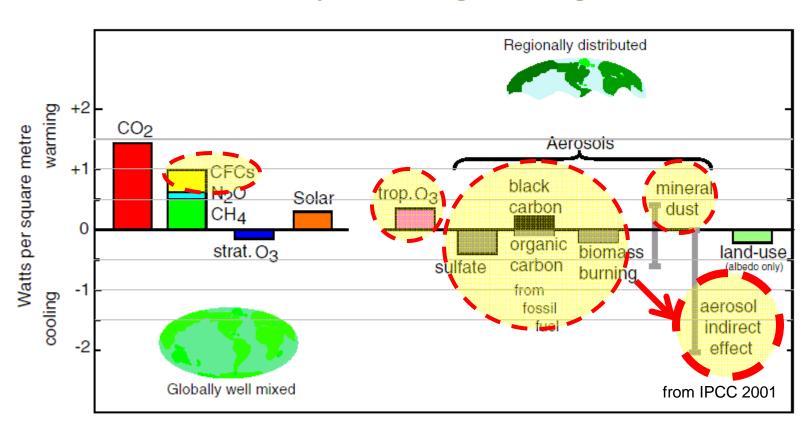
Black (Elemental) Carbon Aerosols

Organic Carbon Aerosols

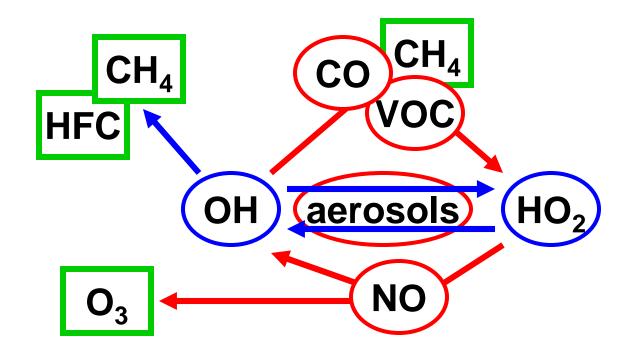
Sulfate & Nitrate Aerosols

Dust (land-use change, agriculture, construction)

Radiative Forcing 1750 → 2000 Non-Kyoto forcings are large!



How do criteria pollutants interact with greenhouse gases?

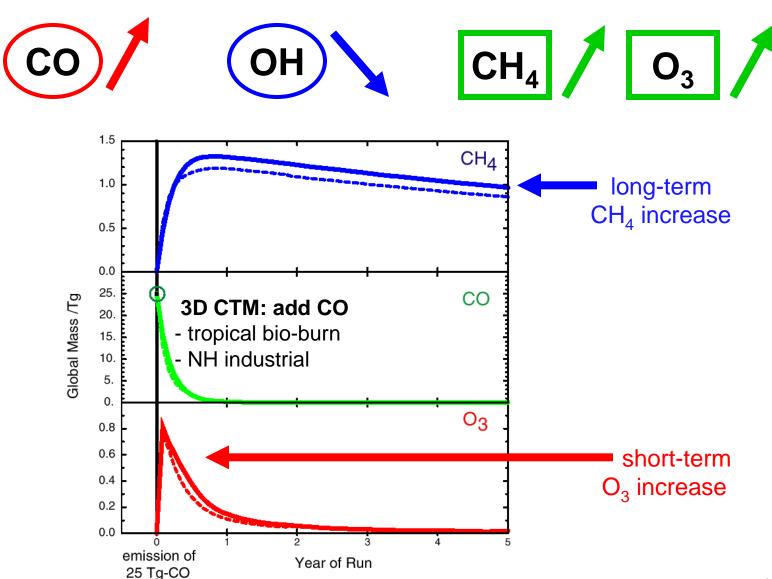


CO, VOC, NO_X (=NO+NO₂), aerosols & CH₄ control

Tropospheric Chemistry (OH, HO₂)

which is the sink for CH₄ & HFCs; the source for O₃

CO becomes an indirect greenhouse gas



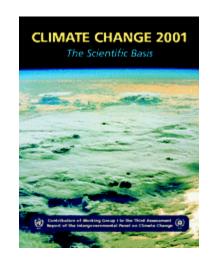
Executive Summary

CO indirect, 2001

Two important new findings since the IPCC WGI Second Assessment Report (IPCC, 1996) (hereafter SAR) demonstrate the importance of atmospheric chemistry in controlling greenhouse gases:

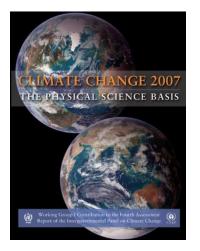
Currently, tropospheric ozone (O_3) is the third most important greenhouse gas after carbon dioxide (CO_2) and methane (CH_4). It is a product of photochemistry, and its future abundance is controlled primarily by emissions of CH_4 , carbon monoxide (CO), nitrogen oxides (NO_x), and volatile organic compounds (VOC). There is now greater confidence in the model assessment of the increase in tropospheric O_3 since the pre-industrial period, which amounts to 30% when globally averaged, as well as the response to future emissions. For scenarios in which the CH_4 abundance doubles and anthropogenic CO and NO_x emissions triple, the tropospheric O_3 abundance is predicted to increase by an additional 50% above today's abundance.

CO is identified as an important indirect greenhouse gas. An addition of CO to the atmosphere perturbs the OH-CH₄-O₃ chemistry. Model calculations indicate that the emission of 100 Mt of CO stimulates an atmospheric chemistry perturbation that is equivalent to direct emission of about 5 Mt of CH₄.



CO indirect GWP

2007



2.10.3.2 Carbon Monoxide

The indirect effects of CO occur through reduced OH levels (leading to enhanced concentrations of CH₄) and enhancement of ozone. The TAR gave a range of 1.0 to 3.0 for the 100-year GWP. Since the TAR, Collins et al. (2002) and Berntsen et al. (2005) have calculated GWPs for CO emissions that range between 1.6 and 2.0, depending on the location of the emissions. Berntsen et al. (2005) found that emissions of CO from Asia had a 25% higher GWP compared to European emissions. Averaging over the TAR values and the new estimates give a mean of 1.9 for the 100-year GWP for CO.

CO and VOCs are easier: both indirect impacts (CH₄ and O₃) are same sign

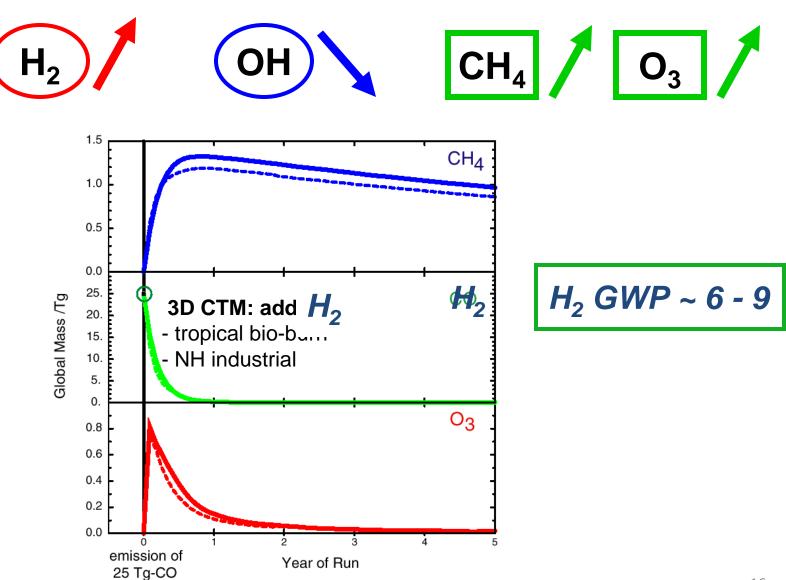
Organic Compound/Study	GWP ^{CH} ₄	GWP ^O 3	GWP
Ethane (C ₂ H ₆)	2.9	2.6	5.5
Propane (C ₃ H ₈)	2.7	0.6	3.3
Butane (C ₄ H ₁₀)	2.3	1.7	4.0
Ethylene (C ₂ H ₄)	1.5	2.2	3.7
Propylene (C ₃ H ₆)	-2.0	3.8	1.8
Toluene (C ₇ H ₈)	0.2	2.5	2.7
Isoprene (C ₅ H ₈)	1.1	1.6	2.7
Methanol (CH ₃ OH)	1.6	1.2	2.8
Acetaldehyde (CH ₃ CHO)	-0.4	1.7	1.3
Acetone (CH ₃ COCH ₃)	0.3	0.2	0.5



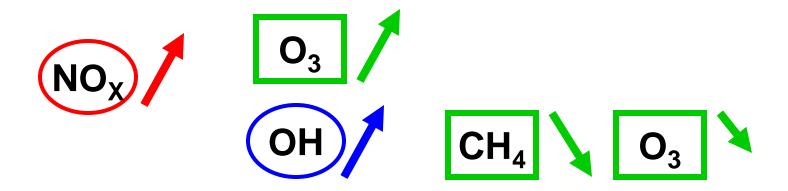


*H*₂ is likewise an indirect greenhouse gas

(Prather, 2003)



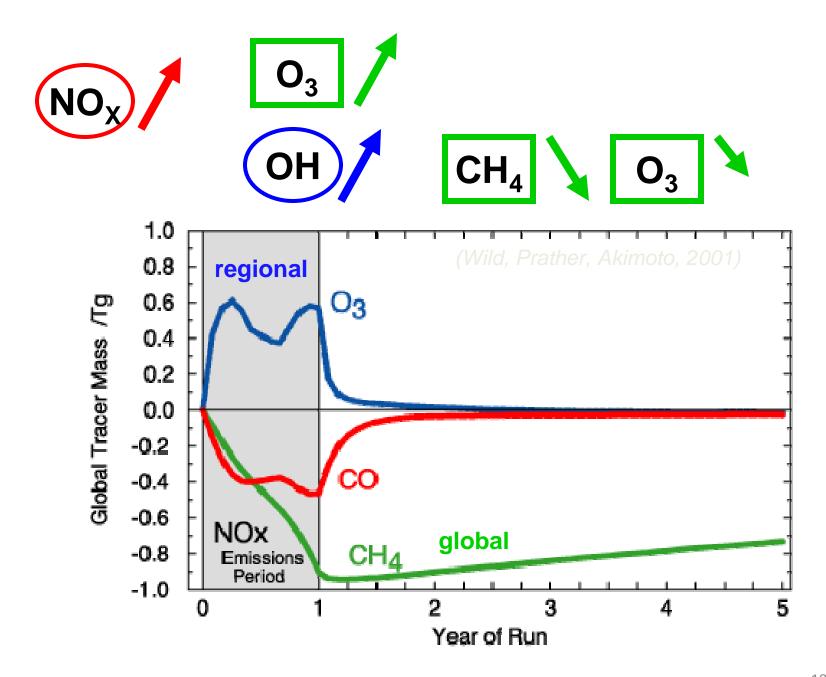
NOx identified as an indirect greenhouse gas



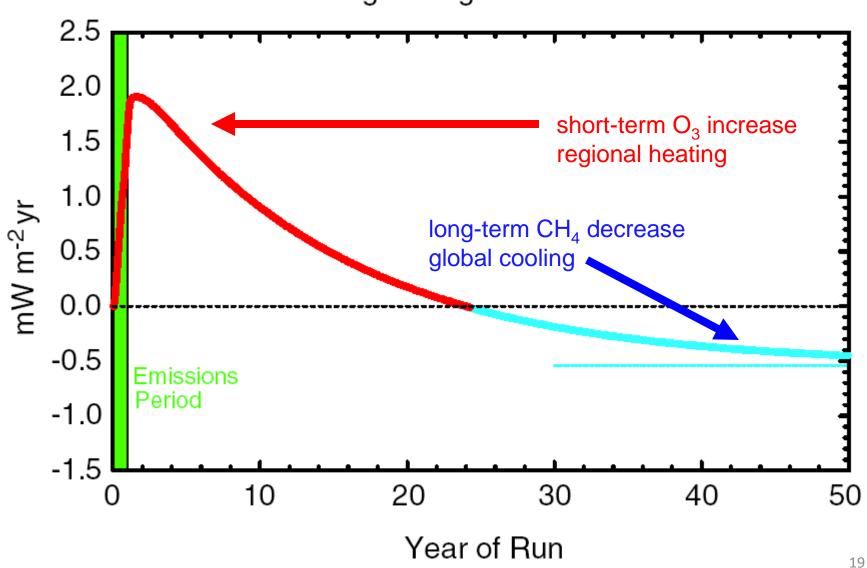
NOx proposed as indirect greenhouse gas (Shine, Derwent, et al, 1st IPCC Assessment Report)

Rejected as too far ahead of its time (1992 IPCC Interim Report)

Post SAR, explicit in 1999 IPCC Aviation Assessment, NOx is indirect greenhouse gas.



Integrated Radiative Forcing (CH₄ & trop O₃) from 0.5 Tg-N as global fossil fuel



NOx is harder:

indirect impacts (CH₄ and O₃) have opposite sign depend greatly on location & season

Organic Compound/Study	GWP ^{CH} ₄	GWP ^o ₃	GWP
Derwent et al. NH surface NO _x a,b	-24	11	-12
Derwent et al. SH surface NO _x a,b	-64	33	-31
Wild et al., industrial NO _x	-44	32	-12
Berntsen et al., surface NO _x Asia	-31 to −42°	55 to 70°	25 to 29°
Berntsen et al., surface NO _x Europe	−8.6 to −11°	8.1 to 12.7	−2.7 to +4.1°
Derwent et al., Aircraft NO _x a,b	-145	246	100
Wild et al., Aircraft NO _x	-210	340	130
Stevenson et al. Aircraft NO _x	-159	155	-3



Aerosols are even more difficult:

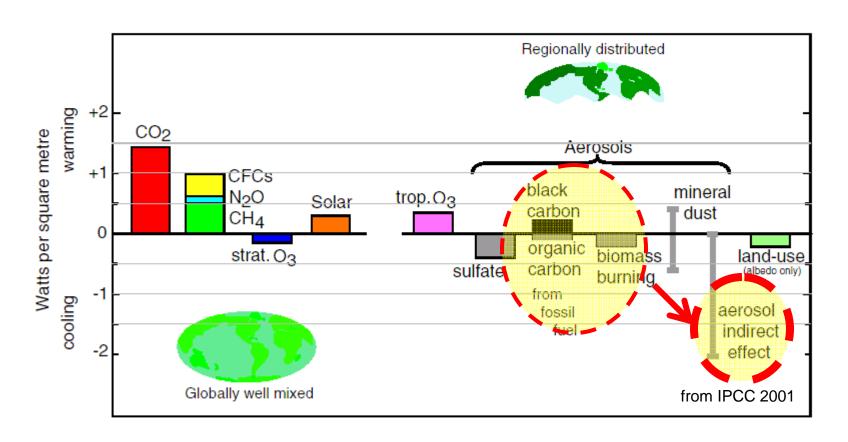
IPCC does not give GWPs for aerosols!

Even more time-space variable than NOx.

Black Carbon (BC) is never emitted alone, and is accompanied by organic carbon & "other" stuff.

Aerosols from fossil fuel and biomass burning (BC + OC) have both heating & cooling

Radiative Forcing 1750 → 2000



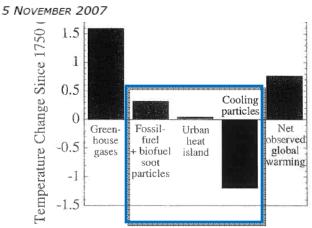
BC aerosol is even more difficult:

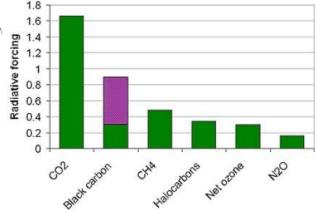
It is not clear that published GWPs for BC aerosols truly represent the full consequence of PM control.

Recent individual publications tend to represent scientific advocacy/viewpoints, not yet assessed

Figure 1. Warming Effect of Various Air Pollutants

Black Carbon May be Second-Most Significant Global Warming Pollutant After Carbon Dioxide; Alters Picturof Diesel Engine Benefits





Green bars indicate IPCC estimates; purple segment indicates findings of recent research. Units are radiative forcings, a measure of influence on global energy exchange, expressed in watts per square meter.

Source: EarthTrends, 2008 using data from IPCC, 2007

ODS emissions may be simpler:

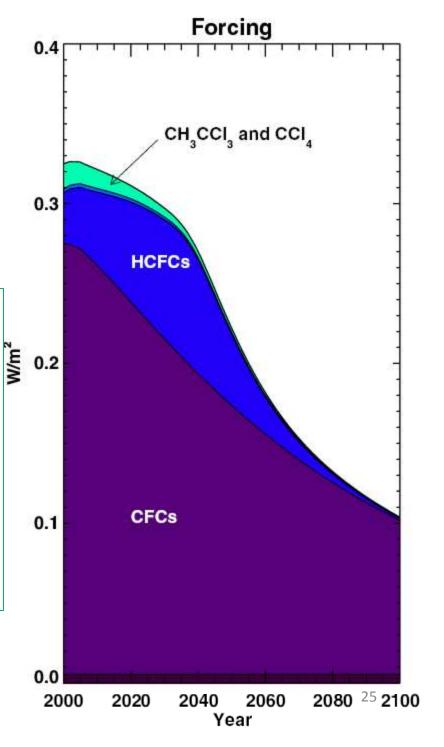
Although phased-out, CFCs & other ODS continue to be emitted.

	Chemical	Use
\Rightarrow	CFC-11	Pre-1994 central A/C, refrigerators
	CFC-12	A/C in pre-1994 motor vehicles
	Halons	Pre-1994 fire suppression systems
	HCFC-141b	Pre-2003 insulation foam in buildings and appliances
*	HCFC-22	Pre-2010 window units, central A/C, commercial refrig.
	HCFC-142b	Pre-2010 foam insulation
	HCFC-225	Pre-2015 solvents
	HCFC-123	Pre-2015 central A/C and fire suppression equipment

ODS emissions continue to drive radiative forcing

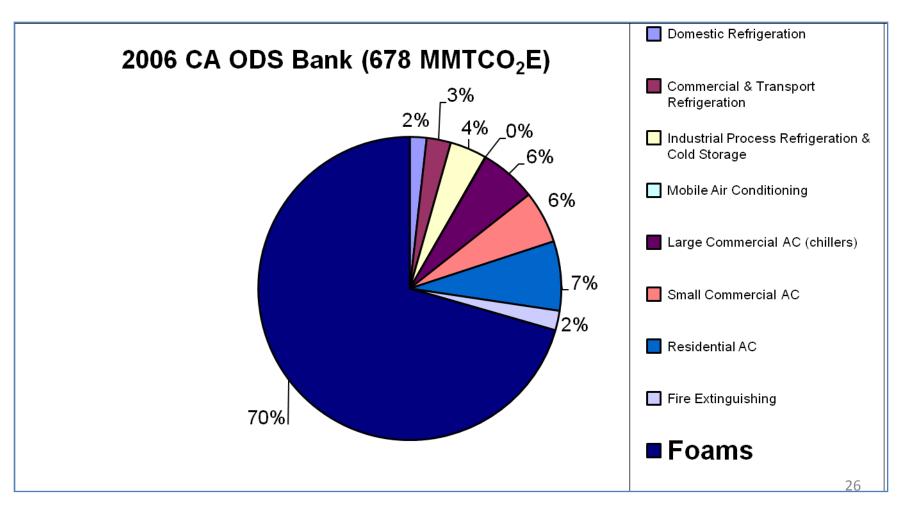
worldwide banks:

Species	WMO (2003) 2002 Bank	IPCC/TEAP (2005) 2002 Bank	This Assessment 2002 Bank
CFC-11	594	1,687	1,654
CFC-12	0	711	711
CFC-113	7	0	5
HCFC-22	1,317	1,531	1,531
HCFC-141b	753	836	674
HCFC-142b	210	224	224
Halon-1211	72	125	125
Halon-1301	58	42	42



ODS emissions may be simpler:

California CFCs are emitted from existing banks, such as insulating foams.



Non-Kyoto Climate Forcers in California

Preliminary Calculations

Climate Forcer	100-year Global Warming Potential ^a	MMT ^b (2005)	MMTCO ₂ E	2005-2020 Change ^c
СО	1.0 – 3.0	4.56	5 – 15	-36%
VOC=ROG	1.1 – 6.2	0.81	1 – 5	-18%
NO _X	Zero to 20	1.07	0 – 20??	-32%
Diesel PM	500 – 1,200??	0.029	15 to 35 minus indirect RF??	-85% ^d
Other PM	unknown		likely negative	
CFC, HCFC	100 — 10,000	0.014	~40	-50% ^e

^a Fossil fuel soot GWP range from Hansen et al. (2007) and Jacobson (2005), all others from IPCC

^b CFC and HCFC estimate from USEPA Vintaging model, all others from CARB emission inventory

[°]CO₂ etc.: AB 32 target. CO, ROG, NO_x: CARB emission inventory for rules already adopted.

^d Diesel PM: 2000 to 2020 Diesel Risk Reduction Plan target.

^e The reduction is based on phase out schedule of the CFCs/HCFCs and does not consider ongoing or planned reduction measures.

Conclusions: How much can California do with non-Kyoto greenhouse gases and aerosols?

Preliminary Calculations

Climate Forcer	MMTCO ₂ E	2005-2020 Change ^a
Kyoto (CO ₂ , CH ₄ , N ₂ O, HFC, PFC, SF ₆)	~500	-15%
Ozone (CO, VOC, NO _X)	zero to 40	-30%
Diesel PM	~25 minus indirect cloud effects ??	-85% b
Other PM	negative	
Montreal (CFC, HCFC)	~40	-50% ^c

^a CO₂ etc.: AB 32 target. CO, ROG, NO_X: CARB emission inventory for rules already adopted.

^b Diesel PM: 2000 to 2020 Diesel Risk Reduction Plan target.

^c The reduction is based on phase out schedule of the CFCs/HCFCs and does not consider ongoing or planned reduction measures.